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number.

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- single atom with a positive or negative charge resulting from the loss or gain of

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electrons, respectively - groups 1A-3A
form cations with positive charge
corresponding to their group number

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Naming Ions & Counting Atoms: Chapter
9.1 Naming & Writing Formulas for Ionic
Compounds: Chapter 9.2 Naming &

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Writing Formulas for Molecular
Compounds: Chapter 9.3 Naming &
Writing Formulas for Acids: Chapter 9.4
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Keys Ionic Compound Naming and
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Chapter 9 Chemical names and formulas. Chemical formula. Empirical formula. molecular formula. functional groups. notation that shows the relative # of atoms in a molecule. a formula that shows the lowest whole # ratio of elements in a.... a formula that shows the true ratio of elements in a compound.

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keep it here for the exam. :D. Terms in this set (7) monoatomic ions.

Chapter 9 Vocabulary: Chemical Names and Formulas ...

Chapter 9 - Chemical Names and Formulas. Jennie L. Borders. Periodic Table for Naming. Section 9.1 Regular Metals Review. The regular metals

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include Group 1 (except H) and 2 and aluminum. When naming a compound that starts with a regular metal, name the first element and add -ide to the second element (except for polyatomic ions).

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Answers

FORMULAS 9. Chapter 9 Chemical Names and Formulas 85 Names and Formulas for Bases (page 273) 6. A base is a compound that produces _____ when dissolved in water. 7. How are bases named?

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Answers

Chapter 9 Chemical Names and Formulas
SECTION 9.3 NAMING AND WRITING FORMULAS FOR MOLECULAR COMPOUNDS (pages 268–270) This section explains the rules for naming and writing formulas for binary molecular compounds. Naming Binary Molecular Compounds (pages 268–269)

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Name Date Class CHEMICAL NAMES AND FORMULAS 9

write the chemical . symbol/name given
the name/chemical symbol of . an .
element. classify . a given compound as
ionic or molecular. g. iven . a sheet of
common ions, represent an ionic
compound in multiple ways--using words
. and chemical formula. represent a

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molecular compound in multiple ways--
using words, chemical formula and
diagrams.

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activities; Start over; Help; Use these activities to help yourself to learn the vocabulary and major topics presented in this chapter. A B; anion: any atom or group of atoms that has a negative charge:

Quia - Chapter 9 "Chemical Names and Formulas"

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Answers

Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer Section MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253 OBJ: 9.1.1 Identify the charges of monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H PTS: 1 DIF: L1 REF: p. 254

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Answers

Chapter 9 Practice Test - Naming and Writing Chemical Formulas

Chemical Names and Formulas 281

CHAPTER 9 Assessment 42 a 2- b 1+ c

1- d 3+ 43 a 2+ b 2+ c 3+ d 1+

44 a barium ion b iodide ion c silver ion d

mercury(II) ion 45 cyanide, CN^- and

hydroxide, OH^- 46 a hydroxide ion b

lead(IV) ion c sulfate ion d oxide ion 47

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zero; A compound is electrically neutral
48 The symbols for the cation and

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Chemical Names And Formulas Chapter
9 Quiz Chemical Names And Formulas
Chapter Unit 5: Chemical Names &
Formulas Chemical Names & Formulas

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(Chapter 9) 2 3 Monatomic Ion
Formation & Nomenclature Monatomic
Ions consist of a _____ atom with a
positive or negative charge resulting
from the gain or loss of one or more
valence _____ Cation: any atom or

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Chapter 9 Chemical Names and Formulas 223 2 Complete this table by writing correct formulas for the compounds formed by combining positive and negative ions 3 Name the following compounds a K_3PO_4 4 c $NaHSO_4$ 4 e N_2O_5 5 g PI_3 3 b $Al(OH)_3$ 3 d HgO f NBr

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Chemical Compounds 9.4 Relationships
Between Masses of Elements and
Compounds 9.5 Determination of
Empirical and Molecular Formulas Write
or recognize the definition of isotope.

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(Section 2.4) Describe the general structure of molecular and ionic compounds. (Sections 3.3 and 3.5) Convert between the names of compounds and their chemical formulas.

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