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17.1: Rates of reactions and rate laws. Chemical change is guided and driven by energetics, but the actual route it takes and the speed with which it occurs is the subject of "dynamics".

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Kinetics • Studies the rate at which a chemical process occurs. • Besides information about the speed at which reactions occur, kinetics also sheds light on the reaction mechanism (exactly how the reaction occurs).

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Reaction dynamics is a field within physical chemistry, studying why chemical reactions occur, how to predict their behavior, and how to control them. It is closely related to chemical kinetics, but is concerned with individual chemical events on atomic length scales and over very brief time periods.

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Chemical kinetics is the study of chemical

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processes and rates of reactions. This includes the analysis of conditions that affect speed of a chemical reaction, understanding reaction mechanisms and transition states, and forming mathematical models to predict and describe a chemical reaction.

What Is Chemical Kinetics?

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17.1: Rates of
reactions and rate laws

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kinetics, is the branch
of physical chemistry
that is concerned with
understanding the
rates of chemical
reactions. It is to be
contrasted with
thermodynamics,

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which deals with the direction in which a process occurs but in itself tells nothing about its rate.

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