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convert uranium oxides to uranium hexafluoride, UF_6 , whose concentration of uranium-235 can ...

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0.5 mol 9. 6.02 $\times 10^{23}$ 10. four moles 11.
6.02 $\times 10^{23}$ Cu atoms 12. 1 mol Cu 13. 4 $\times 10^{23}$
1 mol CH_4 14. 6.02 $\times 10^{23}$ molecules CH_4 15. 23 1

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mol Xe 6.02 $\times 10^{23}$ molecules Xe 14. 23 2 ...

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Chapter 10 Study Guide. How many moles are in 34.6 L of BaO? How many grams are in 4.6 moles of PbSO₄? How many molecules are in 6.5 $\times 10^3$ moles of CO₂? If you have 2.68 moles of S₂O₃, how many atoms of S do you have? How many liters are in 4.39 $\times 10^{22}$ atoms of Mg? The compound methyl butanoate smells like apples.

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